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Synthesis of the C1-C9 Segment of Epothilons

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Abstract: The C1-C9 segment of epothilons was generated by an aldol reaction between chiral aldehyde 3 and ethyl ketone 4. Removal of the TBS protecting groups and debenzylation generated spiro ketal 13 which was analyzed by X-ray crystallography. © 1997 Elsevier Science Ltd. All rights reserved.

Herein we wish to report on the stereoselective synthesis of the C1-C9 fragment of epothilons (Scheme 1). The epothilons A und B (1) were isolated by Höfle et al.^{1,2} and are considered promising anti-tumor agents since they are more active than taxol.³ In our retrosynthetic analysis of epothilons the connection of two fragments can be made between C9 and C10. The substructure C1-C9 contains four of the seven asymmetric centres and should be easily synthesized by an aldol reaction of the key intermediates 3 and 4 (Scheme 2). A similar C6-C7 dissection has already been employed by Schinzer et al.⁴ Nicolaou et al.⁵ and Mulzer et al.⁶ in their retrosynthetic analysis of epothilons.



Gennari, Paterson et al.⁷ demonstrated that aldol reactions of Z-enolates with aldehyde 3 generate the desired 6,7-syn-7,8-anti configuration in the major product. However, the influence of the asymmetric center at C3 on the stereochemistry of the aldol reaction was difficult to assess. In this aldol reaction the conformation of the enolate or a Cram chelation-controlled transition state could establish the desired stereochemistry. Nevertheless, we expected the enolate geometry to govern the stereochemistry of the aldol product. Fragment 4 was synthesized starting from diol 5 (Scheme 3).



Scheme 2

Benzylation followed by Swern oxidation generated aldehyde 6. Wittig-Horner olefination and successive reduction with DIBAI-H gave allyl alcohol 8, which was subjected to a Sharpless epoxidation to generate epoxide 9.⁸ Opening the epoxide with Red-Al gave the 1,3-diol,⁹ which was used to determine the enantiomeric excess by ¹H NMR shift experiments with $(+)Eu(hfc)_3^{10}$ (93% ee). Protecting with TBSCl,¹¹ hydrogenation of the benzyl protecting group and Swern oxidation gave aldehyde 11.¹² Addition of EtMgBr followed by oxidation yields building block 4 (Scheme 3).



Scheme 3. a) BnCl, KOt-Bu, dioxane, 2 h, 90 °C, 91 %; b) Swern ox., 1 h, -78 °C \rightarrow 0 °C, 93 %; c) (EtO)₂P(O)CH₂CO₂Et, NaH, toluene/THF, 1 h, 0 °C \rightarrow RT, 93 %; d) DIBAI-H, Et₂O, 3 h, -78 °C \rightarrow RT, 99 %; e) Sharpless: MS 4 Å, Ti(OiPr)₄, (-)-diethyl tartrate, *i*BuOOH, CH₂Cl₂, -40 °C \rightarrow RT, 96 %; f) Red-Al, THF, 1 h, 0 °C, 96 %; g) TBDMSCl, imidazole, DMF, 44 h, 60 °C, 98 %; h) H₂, 10 % Pd/C, EtOH, 48 h, RT, 80 %; i) Swern ox., 1 h, -78 °C \rightarrow 0 °C, 97 %; k) EtMgBr, THF, 0 °C, 15 min, 82 %; l) Jones ox. 0°C, 10 min, 80 %.

Aldehyde 3 was synthesized according to literature procedure,¹³ starting from (S)-(+)-3-hydroxy-2-methyl propionic acid methyl ester by protecting the alcohol as the benzyl ether followed by a reduction / oxidation

sequence. The aldol reaction was performed by reacting the lithium enolate of ethyl ketone 4 with aldehyde 3 and only one aldol product was observed.¹⁴ X-ray crystallography was applied to determine the configuration of the new asymmetric centers. Since none of the protected derivatives of 2 gave single crystals the TBS protective groups were removed by treatment with HF and the benzyl protective group was removed by hydrogenation with H₂ Pd/C. Under hydrogenation conditions triol 12 cyclized to spiro ketal 13, of which a single-crystal X-ray structure¹⁵ was obtained. (Scheme 4).



Scheme 4. Aldol reaction and synthesis of spiro ketal 13: a) LDA, THF, 15 min, -78 °C, 64 %; b) HF (48 %), MeCN, 3 h, RT, 69 %; c) H₂, Pd/C, 24 h, RT, 86 %.

Figure 1 shows the X-ray structure of spiro ketal 13. We have thus demonstrated that the configuration of C3 can be incorporated in the ketone fragment of the epothilon precursor and that upon aldol reaction the C1-C9 segment can be generated with the desired stereochemistry.



Figure 1. Molecular structure of compound 13.

References and Notes

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- 14. IR (Film): v = 3028, 2956, 2928, 2856, 1684, 1472, 1256, 1100. ¹H-NMR (400 MHz, CDCl₃): $\delta = 0.02$ (s, 6 H, SiMe₃), 0.06 (s, 3 H, SiMe₃), 0.09 (s, 3 H, SiMe₃), 0.87 (s, 9 H, C(CH₃)₃), 0.88 (s, 9 H, C(CH₃)₃), 0.95 (d, 3 H, J = 6.8 Hz, CHCH₃), 1.04 (d, 3 H, J = 6.8 Hz, CHCH₃), 1.07 (s, 3 H, CH₃), 1.19 (s, 3 H, CH₃), 1.42-1.67 (m, 2 H, -OCH-CH₂-CH₂O), 1.81-1.90 (m, 1 H, CH-CH₃), 3.21-3.28 (m, 1 H), 3.42-3.49 (m, 2 H), 3.49-3.61 (m, 2 H), 3.61-3.68 (m, 1 H), 3.88 (dd, 1 H, J = 2.7 Hz, J = 7.5 Hz, -OCH-CH₂-CH₂O-), 4.5 (d, 1 H, J = 2.6, Ph-CH₂-O-), 7.20-7.33 (m, 5 H, arom.); ¹³C-NMR (100 MHz, CDCl₃): $\delta = -5.3$ (+, SiCH₃), -4.1 (+, SiCH₃), -3.8 (+, SiCH₃), 9.7 (+, CH₃), 14.0 (+, CH₃); 18.2 (Cq, C(CH₃)₃), 20.3 (+, CH₃), 23.0 (+, CH₃), 25.9 (+, C(CH₃)₃), 26.0 (+, C(CH₃)₃), 36.3 (+), 37.8 (-), 41.5 (+), 53.9 (Cq), 60.4 (-), 72.7 (+), 72.9 (-), 73.2 (-), 74.1 (+), 127.4 (+), 127.5 (+), 128.2 (+), 128.8 (+), 138.6 (Cq), 221.6 (Cq, Keton).
- 15. Crystal data for 13: $C_{13}H_{24}O_4$, CSD 406240, orthorhombic, $P_{2_1}2_{1_2}1_1$, T = -100 °C, a = 705.05 (12), b = 1147.2 (2), c = 1572.5 (2) pm, Z = 4, $D_C = 1.276$ Mg/m³, $\lambda = 71.073$ pm, $\mu = 0.093$ mm⁻¹, crystal size 0.4 x 0.35 x 0.3 mm, θ range for data collection = 3.14 to 27.50 °, reflections collected 1908, independent 1896, data / restraints / parameters 1896 / 155 / 160, wR(F2) = 0.082, R(F) = 0.041, goodness-of-fit on F² = 0.898, largest diff. peak and hole 204 and -176 e.nm⁻³. Other details can be obtained from the Fachinformationszentrum Karlsruhe, 76344 Eggenstein-Leopoldshafen, Germany. Any request for material should quote a full literature citation and the reference number CSD 406240. The absolute configuration could not be crystallographically established.

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